## **Practice Exam 3i - Answers**

- 1. D, F, J, O, Q, V, AA, CC
- 2. (a) see Solomons Figure 6.7

  - (d) The carbocation which would be formed by dissociate of bromide anion cannot attain planarity (i.e., to become  $sp^2$ )
- 3. (a) The conjugate base (anion) of **A** is resonance stablilized, whereas ethoxide, the conjugate base of ethanol, is not.

(b) 
$$H = Br$$
  $H_3C = OH_2 : CI = OH_3CH_2 : C$ 

(d) The pair of electrons of the acetylide anion are in a lower energy sp orbits compared to the pair of electrons of the CH<sub>3</sub>CH<sub>2</sub>: anion (sp<sup>3</sup> orbital)